Gold(I)-Catalyzed Cycloisomerization−Dimerization Cascade of Benzene-Tethered 1,6-Enynes

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S [Supporting Information](#page-7-0)

ABSTRACT: An unprecedented stereoselective domino reaction of 1,6 enynes with an aryl ring at C3−C4 in the presence of gold(I) catalysts at low temperature is described. This process involves a novel 5-exo-dig cycloisomerization−dimerization sequence to afford formal Diels−Alder adducts that undergo a smooth gold-catalyzed double bond migration at room temperature. In addition, the first examples of the gold mesoionic carbene mediated $[2+2+2]$ cycloaddition of these enynes with benzaldehyde are reported.

ENTRODUCTION

The highly versatile reactivity of 1,6-enynes in the presence of $\text{gold}(I)$ catalysts is well-established.^{[1](#page-7-0)} The cycloisomerization of these enynes is particularly useful due to the broad array of products potentially accessible via the different intermediates that are currently accepted depending on the reaction conditions and the gold catalysts used. Thus, 1,6-enynes I undergo cycloisomerizations to form cyclopropyl metal carbene intermediates II (5-exo-dig) and/or III (6-endo-dig) (Scheme 1 ,^{[2](#page-7-0)} which evolve through skeletal rearrangements by single cleavage to afford dienes IV if external nucleophiles are absent. 3

Scheme 1. Rationale for Cycloisomerization of 1,6-Enynes

In the presence of nucleophiles, they give rise to substituted exocyclic alkenes V. [4](#page-7-0) Alternatively, intermediates III may lead to fused cyclopropanes VI or to methylenecyclohexenes VII.^{[3](#page-7-0)} Furthermore, gold(I) catalysis encompasses a high potential for the construction of complex target compounds in a domino fashion by trapping intra- or intermolecularly the carbene intermediates with suitable functional groups, producing a rapid increment of molecular complexity in a single operation. In this regard, alkenes can act as partners in cyclopropanations, 5 and carbonyl derivatives may lead to cycloadducts of strikingly different structures depending upon the substitution of the alkene moiety of the enynes. 6 We recently disclosed the straightforward synthesis of gold mesoionic carbenes with a chiral sulfoxide group at C4 and have begun exploring their catalytic activity in the cycloisomerization of 1,6-enynes. These sulfinyl gold carbenes displayed a good catalytic profile in selected examples, producing 1,3-dienes in good yields and with high 5-exo regioselectivity that appears to be related to the steric bulk of the N1-substituent of the catalyst.^{[7](#page-8-0)} Seeking to broaden the scope of the process, we examined the reactivity of 1,6-enynes with an aromatic tether such as 1-(ethynyl)-2-(3 mehylbut-2-enyl)benzene 1a that provided a dimeric fluorene-

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Table 1. Cycloisomerization−Dimerization Cascade of 1-(Ethynyl)-2-(3-methylbut-2-enyl)benzene 1a under Gold(I) Catalysis

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 a In all cases, the additive was employed in an amount to equimolar to that of the gold(I) catalyst. b Measured by ${}^1{\rm H}$ NMR of the crude. Unoptimized yields in parentheses. "Yield of the combined dimerization."⁴NaBArF was added in 4 portions at 1 h intervals to produce 3a cleanly; the reaction was allowed to continue for an additional hour (5 h total) to afford pure 3a in high yield after chromatography. ^{*e*}Addition of increasing amounts of NaBArF to this mixture up to 1 equiv led to a 60% yield of 3a after 2 h. ^f TLC and ¹ H NMR analyses showed a complex mixture and substantial decomposition.

like structure assigned as 2a (Table 1, Figure 1). This unprecedented serendipitous finding along with the presence of fluorene-like skeletons in natural products^{[8](#page-8-0)} prompted us to study the process in some depth, and the results are disclosed herein.

RESULTS AND DISCUSSION

Upon examining the reaction of 1-(ethynyl)-2-(3-methylbut-2 enyl)benzene $(1a)$ with catalyst SS-L₁-AuCl $(Table 1)$ and NaBArF (3 mol % each) in CH_2Cl_2 at rt (1 h), we obtained an excellent yield (90%) of an unexpected product 2a (Table 1) that had spectral features (NMR, MS) that were clearly different from those reported for the expected diene 4a. Careful examination of the data for 2a suggested a dimeric structure like the one shown in Table 1 and this was then rigorously established by X-ray diffraction analysis (Figure 1).

A more detailed examination of the original experiment revealed minor amounts of a related product 3a in the crude reaction mixture (Table 1, entry 1). In an effort to clarify the structure of 3a, the process was tested with diastereomeric catalyst $RS-L_1$ -AuCl that afforded comparable results (Table 1, entry 2). Fortunately, careful control of the reaction temperature (−10 °C) led to very clean crude reaction mixtures with catalyst SS-L₁-AuCl (Table 1, entry 3) that produced $3a$ as a single isomer and in excellent yields (91%). At this stage, a

detailed spectroscopic study of 3a, including 2D NMR experiments, strongly supported the proposed structure that may be described as the cycloadduct of a formal Diels−Alder regio- and stereoselective dimerization of 3-indenyl diene 4a.

Figure 1. X-ray structure of 2a.

To gain insight into these processes, we examined the stability of 3a to a hypothetical gold-catalyzed isomerization at rt to afford 2a and, indeed, submitting pure 3a to the catalytic reaction conditions at rt gave an excellent yield of 2a (Table 1, entry 4). This isomerization appears to require gold catalysis because 3a was stable in the presence of stoichiometric NaBArF (see below entry 6). On the other hand, the treatment of indenyl diene 4a prepared as described in the literature 10 10 10 under

Table 2. Scope of the Cycloisomerization−Dimerization Cascade under Gold(I) Catalysis

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Three mol % of gold(I) catalyst and additive was used except for entry 5, where 5% of each was employed. The reactions were not optimized. between the comparison and and the measurement of the crude. Thirteen percent of isomeric 1,5-enyne, (E)-1-(ethynyl)-
Vields in parentheses. "Isomeric ratios in brackets determined by ¹H NMR of the crude. "Thirteen perce 2-(but-1-enyl)benzene, contained in the batch of starting material was recovered unaltered.

the reaction conditions produced a 25:75 mixture of cycloadducts 2a:3a [\(Table 1,](#page-1-0) entry 5). To assess the effect of NaBArF on the cycloaddition, diene 4a was treated with increasing amounts of NaBArF up to stoichiometric and monitored by ¹H NMR to yield cycloadduct 3a (79%, [Table 1](#page-1-0), entry 6). In view of this unexpected and high-yielding NaBArF catalyzed cycloaddition, we examined the stability of enyne 1a with 3% NaBArF that afforded mainly starting material and just 2% ratio ($^1\rm H$ NMR) of indenyl diene 4a [\(Table 1](#page-1-0), entry 7). Interestingly, addition of increasing amounts of NaBArF and monitoring by $^1\mathrm{H}$ NMR revealed measurable increasing amounts of diene 4a and cycloadduct 3a [\(Table 1,](#page-1-0) entry 8) that ultimately led to clean conversion to 3a (60%) upon addition of a full equivalent of NaBArF. Thus, it appears that NaBArF is not just an adequate promoter of the cyclocondensation of diene 4a to cycloadduct 3a, but remarkably, it is also possible to trigger the full cycloisomerization−cycloaddition cascade from enyne 1a. These results are likely to follow a different reaction pathway than upon $\text{gold}(I)$ catalysis. As a final control experiment, enyne 1a was treated with 3 mol % of catalyst $SS-L_1$ -AuCl in the absence of NaBArF to produce a 94:6 mixture of starting material 1a and diene 4a ([Table 1](#page-1-0), entry 9).

At this stage, we decided to examine the behavior of Ph_3PAuCl , a gold (I) complex commonly used in catalysis, in the cycloisomerization−dimerization cascade of enyne 1a, and the results obtained are summarized in [Table 1](#page-1-0), entries 10−13. Activation with NaBArF resulted in a 20:80 mixture of dimer 3a and diene 4a ([Table 1,](#page-1-0) entry 10). In contrast, activation with $AgSbF₆$ led to a good conversion and yield of isomerized dimer 2a ([Table 1,](#page-1-0) entry 11). The use of 5 mol % $AgSbF_6$ essentially

led to recovery of starting material, and 10 mol % $AgSbF_6$ was required to promote some reaction, affording a complex mixture of side products and a low yield of 2a (38%, [Table 1,](#page-1-0) entry 12). Finally, reducing the catalyst load in cold CH_2Cl_2 (3%, 0 °C) gave rise to a good yield of dimer 3a along with some diene 4a [\(Table 1,](#page-1-0) entry 13).

It is worth noting that, from a qualitative standpoint, our carbene catalysts led to slightly faster and cleaner reactions than Ph₃PAuCl. Furthermore, the possibility of having achieved asymmetric induction in the cycloaddition to either 3a or 2a with our enantiopure carbene catalysts led us to evaluate the optical purity of these cycloadducts. This was carried out by chiral HPLC and, to our dismay, the cycloadducts resulted to be virtually racemic in all cases (see [Supporting Information](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b01273/suppl_file/jo7b01273_si_001.pdf) for details).

At this point, we decided to briefly survey the scope of the process by introducing substituents at the aryl moiety of the 1,6-enynes as well as by varying the substitution of the alkene fragment, and the results obtained are summarized in Table 2. Thus, substrate 1b, with a MeO group at the para position to the allyl residue, gave good yields of cycloadducts 2b (69%) and 3b (87%) upon treatment with our carbene catalyst at different temperatures (Table 2, entries 1 and 2). In contrast, regioisomeric substrate 1c at low temperature afforded a 27:73 mixture of 2c and 3c, suggesting a more facile isomerization in this case. Treatment of this mixture with the carbene catalyst at rt gave an excellent yield of 2c (Table 2, entries 3 and 4).

The effect of a disubstituted terminal alkene was examined next, and substrate 1d, upon treatment with $Ph_3PAuCl/$ AgSbF₆, gave a fair yield of naphthalene derivative 5 (Table 2, entry 5) with spectral data identical to that reported in the Table 3. Cycloisomerization-Cross-Cycloaddition Cascade of 1,6-Enynes with Benzaldehyde under Gold(I) Catalysis

"Enyne 1 was added with a syringe pump (1 mL/h) . The best yields were accomplished with 5 equiv of PhCHO. In all cases, the additive was employed in an amount equimolar to that of the gold(I) catalyst. ^bYields in parentheses. CDiastereoisomeric ratio as determined by ¹H NMR of the crude mixture. ^d An additional 50% conversion to 4a and 8% conversion to 3a was shown by the ¹ H NMR spectrum of the crude.

literature.^{[11](#page-8-0)} Finally, treatment of crotyl derivative 1e (5:1, E/Z mixture) either with our gold carbene or with $Ph_3PAuCl/$ $AgSbF₆$ at rt, produced variable mixtures of cycloadducts 2e:3e depending upon the catalyst/additive used ([Table 2,](#page-2-0) entries 6 and 7). Isomerized adduct 2e was isolated as a single isomer by careful chromatography of these mixtures. In contrast, our carbene at low temperature gave a fair yield of 3e as a single isomer after chromatography, along with a less polar fraction that contained unreacted conjugated 1,5-enyne (13%) present in this batch of starting material and a cycloisomerization derived indenyl diene of Z geometry, (Z) -3-(prop-1-enyl)-1Hindene, that appears to be quite unreactive under these conditions.

While 1,6-enynes bearing an aryl ring at C3−C4 are suitable substrates for $[2+2+1]$ annulations with N-hydroxyanilines^{[12](#page-8-0)} and for $[2+2+3]$ cycloadditions with nitrones,^{[13](#page-8-0)} to the best of our knowledge, the related [2+2+2] cycloadditions with aldehydes remain elusive, in sharp contrast with the structurally related 1,7-enynes.^{[6d](#page-8-0)} In fact, it has been mentioned that these 1,6-enynes gave only cycloisomerization instead of cycloaddition to benzaldehyde under certain gold-catalyzed conditions.^{6d} This prompted us to carry out exploratory experiments on these reactions, and the results obtained are summarized in Table 3.

After some experimentation, we found that slow addition of a solution of enyne 1a to a solution of our catalyst, NaBArF and 5 equiv of PhCHO, afforded oxacycle 6a as a single isomer in 70% yield (Table 3, entry 1). In this manner, formation of homocycloaddition derived products (2a, 3a) and diene 4a was minimized. Switching the gold(I) source to Ph_3PAuCl , (Table 3, entry 2) under identical reaction conditions led to poor yields of a separable 60:40 mixture of 6a and diastereomer 7a along with substantial amounts of diene 4a and a small amount of 3a (50 and 8% conversion, respectively). Finally, substrate 1b with a MeO group para to the allyl moiety gave rise to oxacycle 6b as a single isomer in 72% yield under catalysis by our gold carbene (Table 3, entry 3). The structure and relative stereochemistry of compounds 6 were determined by detailed examination of their spectral data including, 2D NMR experiments that indicated a trans relationship for the key proton atoms (H1 and H9a). The optical purity of these

products was evaluated by chiral HPLC that indicated that they were also racemic (see [Supporting Information\)](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b01273/suppl_file/jo7b01273_si_001.pdf).

A rationale to account for the formation and stereochemistry of benzaldehyde cycloadducts 6 is shown in Scheme 2. A gold containing cationic species 9 formed by the accepted $Au(I)$ cyclization of enyne 1a through intermediate 8 is trapped by the aldehyde to form intermediate $Au(I)$ species 11 and 12. These species evolve to products 6 (from 11) and 7 (from 12) by cationic cyclization and deauration. Because the use of Ph₃PAuCl to promote the cyclization leads to a 60:40 mixture of syn-anti isomers 6 and 7 in low yield, the complete diastereoselectivity and enhanced yield found with our sulfinyl gold carbene catalyst may be related with the sulfinyl moiety participating in the stabilization of intermediate 11 to produce syn-isomer 6. This hypothesis requires that the cyclization of 9 and the aldehyde through chair-like transition state 10^{6d} 10^{6d} 10^{6d} has to be reversible $(Scheme 2).^{14}$ $(Scheme 2).^{14}$ $(Scheme 2).^{14}$ A similar stabilizing effect of a

Scheme 2. Plausible Reaction Pathway for the $Gold(I)$ Catalyzed Reaction of 1a with PhCHO

cationic species by the sulfoxide moiety of catalyst $SS-L_1$ -AuCl, which determines in those cases the reactivity or inertia of the catalyst, was reported recently by us.[7](#page-8-0) At this point, we do not have a full understanding of the nature of this stabilizing interaction.

To clarify the reaction pathway of the cyclodimerization process, a crossover cycloisomerization−dimerization experiment using a 50:50 mixture of 1a and 1b as starting material in the presence of $SS-L_1$ -AuCl was carried out ([Scheme 3\)](#page-4-0). The

resulting reaction mixture contained $(^1\mathrm{H~NMR})$ an equimolar mixture of the four possible different cycloadducts (2a:2b:2ab:2ba) arising from self-condensation of 2a and 2b and cross-condensation of 2ab and 2ba. It can be concluded that no substrate discrimination occurs during the cyclization process.

In parallel to the mechanism depicted in [Scheme 2](#page-3-0), cyclization−deauration of intermediate 9 should form 3-indenyl diene 4a (Scheme 4). Intermediate 9 reacts with diene 4a to produce a cyclized cation 14 through transition state 13. The reaction product 3a is formed from 14 following a pathway analogous to the one depicted for aldehydes in [Scheme 2](#page-3-0). Allylic isomerization produced the conjugated olefin 2a. It should be noted that formation of compound 3a is a formal Diels−Alder reaction of indene 4a that did not take place in the absence of Au(I) except for stoichiometric amounts of NaBArF [\(Table 1,](#page-1-0) entry 8). To rule out a possible Au(I) catalyzed Diels−Alder reaction, highly reactive dienophiles such as methyl 3-methyl-2-butenoate and N-phenylmaleimide were added to the reaction mixture. Diels−Alder adducts arising from diene 4a were not formed, leading instead to variable mixtures of 2a, 3a, and 4a. Other less reactive dienophiles in "normal" (1-hexene, 2-methyl-1-phenyl-1-propene, phenylacetylene, 2,3-dimethyl-1,3-butadiene) or "inverse" Diels−Alder reaction (ethyl vinyl ether) produced analogous results. Therefore, formation of compounds 3 parallels the reaction of species 9 with aldehydes. This species may be formed directly from enyne 1a or from indene 4a by $Au(I)$ cycloreversion ([Table 1,](#page-1-0) entry 5). Finally, regarding the

NaBArF promoted cycloisomerization−dimerization of enyne 1a [\(Table 1,](#page-1-0) entries 7 and 8), we observed a similar effect in the Nicholas reaction promoted by a $Au(I)$ species.^{[15](#page-8-0)} It is tempting to speculate that at least the self-cycloaddition of indenyl diene 4a is facilitated by some anionic acceleration due to the mildly basic NaBArF,^{[16](#page-8-0)} though extensive experimentation is needed to propose a reasonable reaction pathway for these reactions.

■ CONCLUSION

In summary, an unprecedented and facile gold(I) catalyzed cycloisomerization−dimerization cascade of 1,6-enynes with a benzene tether at C3−C4 was reported. The scope of this serendipitous finding regarding the enyne substrates and the gold catalyst sources was explored briefly, and an interesting NaBArF catalyzed cycloisomerization−dimerization process was unveiled. In addition, these 1,6-enynes appear to be adequate substrates for a $\text{gold}(I)$ carbene catalyzed process with benzaldehyde, resulting in a highly diastereoselective [2+2+2] cycloaddition that had not been previously reported in the literature. The presence of the sulfinyl ligand clearly influences the selectivity of the cyclization with benzaldehyde.

EXPERIMENTAL SECTION

General Synthetic Methods. Reagents and solvents were handled using standard syringe techniques. Anhydrous solvents were purified by filtration on a solvent purification system (SPS). Crude products were purified by flash chromatography on 230−400 mesh silica gel with distilled solvents. ${}^{1}H$ and ${}^{13}C$ NMR spectra were recorded at 300, 400, or 500 MHz (1 H NMR) and at 100 or 126 MHz (13 C NMR) using CDCl₃ and d_3 -MeCN at 298 K, unless otherwise stated. Chemical shifts are given in ppm. Only the characteristic peaks in the infrared spectra $(\nu \text{ max})$ are quoted. Optical rotations were measured with a 100 mm path cell, at 20 °C and using a sodium lamp. High resolution mass spectra (HRMS) were recorded using an accurate mass QTOF spectrometer. HRMS MALDI/TOF spectra were recorded in a MALDI TOF/TOF spectrometer in positive mode of detection using dithranol as matrix.

Analytical chiral HPLC was performed with detection at 254 nm using a Chiralcel OD-H column.

Enynes 1^{13} 1^{13} 1^{13} and catalysts SS-L₁-AuCl and RS-L₁-AuCl^{[7](#page-8-0)} were synthesized as previously described. $4a^8$ $4a^8$ and 5^{11} 5^{11} 5^{11} have been previously reported, and only their ¹H NMR is given.

General Procedure for the $[2+2+2]$ Cycloisomerization–
dimerization Cascades. A mixture of the gold(I) catalyst and the activating agent (3–5% mol of each one) in 1 mL anhydrous CH_2Cl_2 was prepared in a 2-necked round-bottom flask under argon. A solution of the enyne (1 equiv, 0.10−0.17 mmol) in 1 mL of

Scheme 4. Plausible Reaction Pathway for the Gold(I) Catalyzed Cycloisomerization−Dimerization of 1a

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anhydrous CH_2Cl_2 was then added in one portion at rt (for 2) or at −10 °C (for 3). The reaction was monitored by TLC to assess the consumption of starting material and diene and by $^1\mathrm{H}$ NMR to assess the 2:3 ratio. The reaction was filtered through a short pad of Celite with hexane or pentane and purified by flash chromatography on silica gel after removal of solvents.

2-(1H-Inden-3-yl)-1,1,3,3-tetramethyl-2,3,4,9-tetrahydro-1H-fluorene $(2a)$. Following the general protocol, enyne 1a $(19.3 \text{ mg}, 0.113)$ mmol), SS-L₁-AuCl (1.90 mg, 3.4 μmol, 3 mol %), and NaBArF (3.01 mg, 3.4 μ mol, 3 mol %) were mixed in 2 mL of CH₂Cl₂ under argon. After purification, 17.4 mg of 2a (0.051 mmol, 90%) was obtained [\(Table 1](#page-1-0), entry 1). Alternatively, reaction of RS-L₁-AuCl (2.80 mg, 5.0) μ mol, 3 mol %), NaBArF (4.40 mg, 5.0 μ mol, 3 mol %) and enyne 1a $(28.4 \text{ mg}, 0.167 \text{ mmol})$ in 2 mL of CH₂Cl₂ gave 23.0 mg of 2a $(0.068$ mmol, 81%) after purification [\(Table 1,](#page-1-0) entry 2). Starting from 3a (32.6 mg, 0.096 mmol), SS-L₁-AuCl (1.60 mg, 2.9 μ mol, 3 mol %), and NaBArF (2.55 mg, 2.9 μ mol, 3 mol %) in 2 mL CH₂Cl₂, 30.3 mg of 2a (0.089 mmol, 93%) was obtained ([Table 1](#page-1-0), entry 4). Using Ph₃PAuCl (3.2 mg, 6.4 μ mol, 5 mol %), AgSbF₆ (2.2 mg, 6.4 μ mol, 5 mol %), and enyne 1a (28.4 mg, 0.167 mmol) in 2 mL of CH_2Cl_2 , 19.7 mg of 2a (0.057 mmol, 90%) was isolated [\(Table 1,](#page-1-0) entry 11).

Data for 2a: crystalline solid. Mp 108−110 °C. R_f 0.18 (Hex). ¹H NMR (400 MHz, CDCl₃) δ 7.37 (m, 3H), 7.15 (m, 5H), 6.34 (s, 1H), 3.38 (s, 2H), 3.32 (s, 2H), 3.02 (s, 1H), 2.37 (br s, 2H), 1.23 (s, 3H), 1.09 (s, 3H), 1.01 (s, 3H), 0.84 (s, 3H). ${}^{13}C(^{1}H)NMR$, HSQC, HMBC (100 MHz, CDCl₃) δ 148.9, 148.6, 146.4, 143.3, 143.2, 143.1, 133.4 (C), 130.4, 126.3, 126.1, 124.5, 124.1, 123.7, 123.6, 119.2, 118.1, 51.7 (CH), 39.1, 38.5 (CH₂), 37.5 (C), 36.5 (CH₂), 35.1 (C), 32.4, 31.2, 25.2, 24.1 (CH₃). IR (KBr): 3042, 2960, 2888, 1629, 1605, 1470, 1458, 1393, 1380, 1362, 1260, 1210, 1019, 972, 778, 756, 719 cm[−]¹ . HRMS (ESI) m/z calcd for $C_{26}H_{28}$ [M]⁺: 340.2186; found 340.2185. Elemental analysis calcd for $C_{26}H_{28}$: C, 91.71; H, 8.29; found: 91.65; H, 8.52. $[\alpha]_D^{25} \sim 0$ ($c = 1.0$, CHCl₃).

6-Methoxy-2-(5-methoxy-1H-inden-3-yl)-1,1,3,3-tetramethyl-2,3,4,9-tetrahydro-1H-fluorene (2b). Following the general protocol, SS-L₁-AuCl (1.73 mg, 3.1 μ mol, 3 mol %), NaBArF (2.75 mg, 3.1) μ mol, 3 mol %), and enyne 1b (20.7 mg, 0.103 mmol) were mixed in 2 mL of CH_2Cl_2 to give 14.4 mg of 2b (0.036 mmol, 69%) after purification (Hex: CH_2Cl_2) ([Table 2](#page-2-0), entry 1).

Data for 2b: crystalline solid. Mp 113-115 °C. R_f 0.18 $(Hex:CH_2Cl_2, 1:1).$ ¹H NMR, COSY, NOESY (400 MHz, CDCl₃) δ 7.30 (d, J = 8.1 Hz, 1H, H7'), 7.24 (d, J = 8.1 Hz, 1H, H8), 6.89 (d, J $= 2.0$ Hz, 1H, H4'), 6.76 (d, J = 2.2 Hz, 1H, H5), 6.70 (dd, J = 8.1, 2.0 Hz, 1H, H6′), 6.65 (dd, J = 8.1, 2.2 Hz, 1H, H7), 6.38 (s, 1H, H2′), 3.79 (s, 3H, OMe), 3.77 (s, 3H, OMe), 3.34 (s, 2H, H1′), 3.29 (s, 2H, H9), 2.95 (s, 1H, H2), 2.35 (s, 2H, H4), 1.23 (s, 3H), 1.10 (s, 3H), 1.04 (s, 3H), 0.86 (s, 3H). 13C{1 H}NMR, HSQC, HMBC (100 MHz, CDCl3) δ 159.2, 159.0, 150.5, 150.2, 147.9, 143.0, 135.5, 135.3, 133.3 (C), 131.9 (C2′), 123.93, 123.89, 109.7 (2), 105.7, 104.2 (CH), 55.7 (2 CH_3) , 51.8 (CH, C2), 39.1 (C4), 37.8 (CH₂, C1'), 37.6 (C), 35.8 $(CH_2, C9)$, 35.1 (C), 32.5, 31.2, 25.2, 24.1 (CH₃). IR (KBr): 2958, 2885, 1618, 1607, 1475, 1362, 1286, 1214, 1160, 1032, 798 cm⁻¹. . HRMS (ESI) m/z calcd for $C_{28}H_{33}O_2$ [M + H]⁺ 401.2475; found 401.2493.

7-Methoxy-2-(6-methoxy-1H-inden-3-yl)-1,1,3,3-tetramethyl-2,3,4,9-tetrahydro-1H-fluorene (2c). Following the general protocol, SS-L₁-AuCl (1.86 mg, 3.3 μ mol, 3 mol %) was dissolved in CH₂Cl₂ under argon at -10 °C. NaBArF (2.96 mg, 3.3 μ mol, 3 mol %) and 1c (22.3 mg, 0.111 mmol) in 1 mL CH₂Cl₂ were subsequently added. After 1 h, the reaction was filtered through Celite using hexane and analyzed by ¹H NMR, which indicated a $2c:3c$ 27:73 ratio ([Table 2,](#page-2-0) entry 3). This mixture was redissolved in 1 mL of CH_2Cl_2 and added onto a mixture of SS-L₁-AuCl (1.86 mg, 3.3 μ mol, 3 mol %) and NaBArF (2.96 mg, 3.3 μ mol, 3 mol %) in 1 mL of CH₂Cl₂ at rt. After 1 h, filtration through Celite (hexane) and flash chromatography $(Hex:CH₂Cl₂), 22.2$ mg of 2c $(0.055 \text{ mmol}, 98%)$ were obtained [\(Table 2,](#page-2-0) entry 4).

Data for 2c: crystalline solid. Mp 162−164 °C. R_f 0.17 $(Hex:CH₂Cl₂, 3:2).$ ¹H NMR, COSY, NOESY (400 MHz, CDCl₃) δ 7.23 (d, J = 8.3 Hz, 1H, H4'), 7.07 (d, J = 8.3 Hz, 1H, H5), 7.03 (d, J $= 1.8$ Hz, 1H, H7'), 7.00 (d, J = 2.3 Hz, 1H, H8), 6.80 (d, J = 8.3 Hz, 2H, H6+H5′), 6.21 (s, 1H, H2′), 3.78 (s, 6H, 2OMe), 3.37 (s, 2H, H1′), 3.32 (s, 2H, H9), 2.96 (s, 1H, H2), 2.34 (s, 2H, H4), 1.23 (s, 3H), 1.10 (s, 3H), 1.02 (s, 3H), 0.86 (s, 3H). 13C{1 H}NMR, HSQC, HMBC (100 MHz, CDCl₃) δ 157.8, 157.6, 146.3, 145.04, 145.01, 142.7, 142.3, 139.7, 132.8 (C), 128.1 (C2′), 119.5, 118.2, 111.6, 111.5, 110.8, 110.2 (CH), 55.9, 55.7 (CH₃), 51.9 (CH, C2), 39.2 (CH₂, C4), 38.4 (CH₂, C1'), 37.4 (C), 36.5 (CH₂, C9), 35.1 (C), 32.4, 31.2, 25.1, 24.1 (CH₃). IR (KBr): 2960, 2925, 2875, 1608, 1595, 1482, 1467, 1381, 1283, 1240, 1225, 1121, 1027, 853, 815, 803 cm⁻¹. HRMS (ESI) m/z calcd for $C_{28}H_{33}O_2$ [M + H]⁺ 401.2475; found 401.2475. Elemental analysis calcd for $C_{28}H_{32}O_2$: C, 83.96; H, 8.05; found: 83.64; H, 8.35.

(1RS,2SR,3RS)-2-(1H-Inden-3-yl)-1,3-dimethyl-2,3,4,9-tetrahydro-1H-fluorene (2e). Following the general protocol, $SS-L_1$ -AuCl (2.29 mg, 4.1 μmol, 3 mol %), NaBArF (3.64 mg, 4.1 μmol, 3 mol %), and enyne 1e (21.4 mg, 0.137 mmol) in 2 mL of CH_2Cl_2 were mixed at rt. After 1 h and filtration through Celite, ¹H NMR indicated a 50:50 mixture of 2e:3e ([Table 2,](#page-2-0) entry 6). Purification (hexane) gave 16.0 mg of the mixture of products (0.051 mmol, 74%). Alternatively, Ph₃PAuCl (2.24 mg, 4.4 μ mol, 3 mol %), AgSbF₆ (1.5 mg, 4.4 μ mol, 3 mol %), and enyne 1e (21.0 mg, 0.134 mmol) were mixed in 2 mL of CH_2Cl_2 and protected from light. Analysis by ¹H NMR after 1 h and filtration through Celite indicated a 72:28 mixture of 2e:3e [\(Table 2,](#page-2-0) entry 7). Purification (hexane) gave 15.3 mg of combined products (0.049 mmol, 73%). After a second careful chromatography, pure 2e (5 mg) was isolated for characterization.

Data for $2e$: colorless oil. R_f 0.17 (Hex). ¹H NMR, COSY (500 MHz, CDCl₃) δ 7.50 (d, J = 7.3 Hz, 1H), 7.47 (d, J = 7.6 Hz, 1H), 7.43 (d, J = 7.3 Hz, 1H), 7.24–7.32 (m, 3H), 7.21 (t, J = 7.3 Hz, 1H), 7.15 (t, J = 7.3 Hz, 1H), 6.30 (t, J = 1.9 Hz, 1H, H2′), 3.46–3.38 (m, 3H, H1' + H9 α), 3.25 (d, J = 22.1 Hz, 1H, H9 β), 3.00 (br s, 1H, H1), 2.73 (d, J = 15.7 Hz, 1H, H4 α), 2.39 (t, J = 10.2 Hz, 1H, H2), 2.28– 2.42 (br s, 1H, H3), 2.18 (m, 1H, H4 β), 1.07 (d, J = 7.0 Hz, 3H), 0.91 (d, J = 6.0 Hz, 3H). ¹H homodecoupling (500 MHz, 318 K): $J_{\text{H1-H2}}$ = 8.4, J_{H2-H3} = 10.8 Hz. NOESY 2D (400 MHz, 328 K): correlation peaks between H2-Me1, H2-Me3, H1-H3, H2′-H3. ¹³C{¹H}NMR, HSQC, HMBC (100 MHz, CDCl₃, 328 K) δ 147.4 (br), 146.0, 145.4, 145.2, 143.4, 135.7 (C), 129.6 (br, C2′), 126.4, 126.0, 124.7, 124.1, 124.1, 123.6, 120.2, 118.1, 50.9 (br, CH, C2), 38.7 (C9), 37.9 (CH₂, C1'), 37.1 (br, C1), 33.8 (br, CH, C3), 32.0 (CH₂, C4), 20.7, 19.6 (CH₃). IR (film): 3019, 2955, 2924, 2854, 1742, 1683, 1457, 1377, 1261, 1154, 968, 769, 719 cm⁻¹. HRMS (MALDI/TOF) m/z calcd for $C_{24}H_{23}$ [M + H]⁺ 311.1799; found 311.1785.

(2SR,9aSR)-2-(1H-Inden-3-yl)-1,1,3,3-tetramethyl-2,3,9,9a-tetrahydro-1H-fluorene (3a). Following the general protocol at −10 °C, SS-L₁-AuCl (1.95 mg, 3.5 μ mol, 3 mol %), NaBArF (3.09 mg, 3.5 μ mol, 3 mol %), and 1a (19.8 mg, 0.116 mmol) in 2 mL of CH₂Cl₂ yielded after purification (Hex) 18.0 mg of 3a (0.053 mmol, 91%) [\(Table 1,](#page-1-0) entry 3). Alternatively, reaction of 4a (10.2 mg, 0.060 mmol) with NaBArF (53 mg, 0.060 mmol, 1 equiv) in 1 mL anhydrous $CH₂Cl₂$ afforded 8.17 mg (0.024 mmol, 79%) of 3a [\(Table 1,](#page-1-0) entry 6). Also, reaction of 1a (22.3 mg, 0.131 mmol) with NaBArF (116 mg, 0.060 mmol, 1 equiv) in 2 mL of CH_2Cl_2 afforded after 2 h 13.4 mg (0.039 mmol, 60%) of 3a [\(Table 1,](#page-1-0) entries 7 and 8).

Data for 3a: crystalline solid. Mp 133–135 °C. R_f 0.17 (Hex). ¹H NMR, COSY, NOESΥ (500 MHz, CDCl₃) δ 7.52–7.42 (m, 3H), 7.32 $(t, J = 7.5 \text{ Hz}, 1\text{H})$, 7.29–7.26 (m, 1H), 7.24–7.16 (m, 3H), 6.49 (t, J $= 2.0$ Hz, 1H, H2'), 5.96 (d, J = 2.5 Hz, 1H, H4), 3.48 (dd, J = 4.1, 1.9 Hz, 2H, H1′), 3.08 (s, 1H, H2), 3.02−2.91 (m, 2H, H9a + H9α), 2.83 (dd, J = 14.4, 7.3 Hz, 1H, H9 β), 1.20 (s, 3H), 1.08 (s, 3H), 1.00 (s, 3H), 0.81 (s, 3H). ¹H NMR, COSY, NOESY (500 MHz, d_3 -MeCN) δ 7.53−7.48 (m, 2H), 7.48−7.43 (m, 1H), 7.33−7.24 (m, 2H), 7.23− 7.15 (m, 3H), 6.54 (t, $J = 2.3$ Hz, 1H, H2'), 6.01 (d, $J = 2.9$ Hz, 1H, H4), 3.48 (s, 2H, H1'), 3.14 (s, 1H, H2), 3.02 (td, J = 8.6, 2.9 Hz, 1H, H9a), 2.95 (dd, J = 15.8, 8.9 Hz, 1H, H9 α), 2.80 (dd, J = 15.8, 8.5 Hz, 1H, H9β), 1.20 (s, 3H), 1.06 (s, 3H), 0.95 (s, 3H), 0.78 (s, 3H). NOESY 2D (500 MHz, d_3 -MeCN): correlation peaks between H2-H9a, H2-Me (0.78), H9a-Me (0.78). ¹³C{¹H}NMR, HSQC, HMBC $(126 \text{ MHz}, \text{CDCl}_3)$ δ 149.0, 144.7, 143.3, 142.6, 141.2, 139.1 (C),

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130.8 (C2′), 127.9 (C4), 127.7, 126.6, 126.1, 125.4, 124.4, 123.6, 120.3, 119.2, 53.7 (C9a), 52.8 (CH, C2), 38.6 (CH₂, C1'), 37.33, 37.25 (C), 33.0 (CH₃), 32.2 (CH₂, C9), 28.1, 26.7, 17.6 (CH₃). 37.25 (C), 33.0 (CH₃), 32.2 (CH₂, C9), 28.1, 26.7, 17.6 (CH₃).
¹³C{¹H}NMR, HSQC, HMBC (126 MHz, d₃-MeCN) δ 149.9, 145.6, 144.3, 143.1, 141.9 140.2 (C), 132.3 (C2′), 128.7, 128.4 (C4), 127.5, 126.9, 126.3, 125.3, 124.6, 121.0, 120.0, 54.1 (C9a), 53.2 (CH, C2), 39.1 (CH₂, C1'), 37.9, 37.8 (C), 32.9 (CH₃), 32.6 (CH₂, C9), 28.1, 27.0, 17.8 (CH₃). IR (KBr): 2952, 2877, 1604, 1461, 1362, 1166, 975, 860, 755 cm⁻¹. HRMS (ESI) m/z calcd for C₂₆H₂₉ [M + H]⁺ 341.2264; found 341.2269. Elemental analysis calcd for $C_{26}H_{28}$: C, 91.71; H, 8.29; found: 91.81; H, 8.60. $[\alpha]_{D}^{25} \sim 0$ ($c = 0.64$, CHCl₃).

(2SR,9aSR)-6-Methoxy-2-(5-methoxy-1H-inden-3-yl)-1,1,3,3-tetramethyl-2,3,9,9a-tetrahydro-1H-fluorene (3b). Following the general protocol, SS-L₁-AuCl (1.85 mg, 3.3 μ mol, 3 mol %), NaBArF $(2.93 \text{ mg}, 3.3 \mu \text{mol}, 3 \text{ mol} \%)$ and enyne 1b $(22.1 \text{ mg}, 0.110 \text{ mmol})$ were mixed in 2 mL of CH_2Cl_2 under argon at -10 °C. After purification (Hex: CH_2Cl_2), 19.3 mg of 3b (0.048 mmol, 87%) was isolated [\(Table 2,](#page-2-0) entry 2).

Data for 3b: crystalline solid. Mp 152-154 °C. R_f 0.23 $(Hex:CH_2Cl_2, 3:2).$ ¹H NMR, COSY, NOESY (400 MHz, CDCl₃) δ 7.31 (d, J = 8.1 Hz, 1H, H7'), 7.09 (d, J = 8.2 Hz, 1H, H8), 6.92 (s, 2H, H5 + H4′), 6.71 (m, 2H, H7 + H6′), 6.44 (s, 1H, H2′), 5.86 (d, J = 2.8 Hz, 1H, H4), 3.77 (s, 6H, 2OMe), 3.35 (s, 2H, H1′), 2.96−2.88 $(m, 2H, H9a + H2)$, 2.81 (dd, J = 15.4, 8.7 Hz, 1H, H9 α), 2.68 (dd, J = 15.3, 8.3 Hz, 1H, H9β), 1.12 (s, 3H), 1.00 (s, 3H), 0.94 (s, 3H), 0.74 (s, 3H). ¹³C{¹H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 159.1, 159.0, 150.7, 142.5, 142.4, 139.4, 136.9, 135.5 (C), 132.4 (C2′), 127.9 $(C4)$, 126.0, 123.9, 115.0, 109.6, 105.7, 104.5 (CH), 55.7, 55.6 (CH₃), 54.3 (C9a), 53.0 (CH, C2), 37.9 (CH₂, C1'), 37.4, 37.3 (C), 33.0 (CH_3) , 31.4 (CH₂, C9), 28.1, 26.8, 17.5 (CH₃). IR (KBr): 2954, 2832, 1607, 1487, 1475, 1362, 1288, 1217, 1158, 1034, 841 cm[−]¹ . HRMS (ESI) m/z calcd for $C_{28}H_{33}O_2$ [M + H]⁺ 401.2475; found 401.2493. Elemental analysis calcd for $C_{28}H_{32}O_2$: C, 83.96; H, 8.05; found: 83.82; H, 8.08.

(2SR,9aSR)-7-Methoxy-2-(6-methoxy-1H-inden-3-yl)-1,1,3,3-tetramethyl-2,3,9,9a-tetrahydro-1H-fluorene (3c). Following the general protocol, SS-L₁-AuCl (1.86 mg, 3.3 μ mol, 3 mol %), NaBArF (2.96 mg, 3.3 μmol, 3 mol %), and enyne 1c (22.3 mg, 0.111 mmol) reacted in 2 mL of CH₂Cl₂ at −10 °C. Purification (Hex:CH₂Cl₂) gave 6.9 mg of 3c (0.017 mmol, 30%), which readily isomerized to a 27:73 mixture of 2c:3c [\(Table 2](#page-2-0), entry 3).

Partial data for 3c from a 27:73 2c:3c mixture: R_f 0.18 $(Hex:CH_2Cl_2, 3:2).$ ¹H NMR, COSY, NOESY (400 MHz, CDCl₃) δ 7.30 (d, J = 8.3 Hz, 1H, H4' or H5), 7.23 (d, J = 8.3 Hz, 1H, H5 or H4′), 7.02 (d, J = 2.3 Hz, 1H, H7′ or H8), 6.79 (dd, J = 8.3, 2.3 Hz, 1H, H5′ or H6), 6.74 (s, 1H, H8 or H7′), 6.71 (dd, J = 8.4, 2.1 Hz, 1H, H6 or H5'), 6.25 (s, 1H, H2'), 5.73 (d, J = 2.8 Hz, 1H, H4), 3.77 (s, 3H, OMe), 3.75 (s, 3H, OMe), 3.37 (s, 2H, H1′), 2.92 (s, 1H, H2), 2.92−2.68 (m, 3H, H9a + H9α+ H9β), 1.10 (s, 3H), 0.99 (s, 3H), 0.91 (s, 3H), 0.73 (s, 3H). 13C{1 H}NMR, HSQC, HMBC (100 MHz, CDCl3) δ 159.9, 157.8, 146.3, 145.0, 142.4, 142.2, 138.5, 134.2 (C), 128.5 (C2′), 125.9 (C4), 121.1, 119.5, 113.2, 111.6, 110.3, 110.2 (CH), 55.7, 55.6 (CH₃), 54.1 (C9a), 53.0 (CH, C2), 38.5 (CH₂, C1'), 37.4, 37.2 (C), 33.1 (CH₃), 32.4 (CH₂, C9), 28.2, 26.9, 17.5 (CH₃). IR (film): 2926, 2855, 1741, 1683, 1482, 1466, 1381, 1289, 1244, 1226, 1141, 1111, 1033, 840, 805 cm[−]¹ . HRMS (ESI) m/z calcd for $C_{28}H_{33}O_2$ [M + H]⁺ 401.2475; found 401.2475.

(1SR,2SR,3RS,9aSR)-2-(1H-Inden-3-yl)-1,3-dimethyl-2,3,9,9a-tetrahydro-1H-fluorene (3e). Following the general protocol, $SS-L_1$ -AuCl (2.31 mg, 4.1 μmol, 3 mol %), NaBArF (3.67 mg, 4.1 μmol, 3 mol %), and enyne 1e (24.8 mg, 87% purity, 0.138 mmol) in 2 mL of CH₂Cl₂ were mixed at -10 °C. Filtration through Celite and purification (hexane) gave 8.90 mg of 3e (0.029 mmol, 48%) and 2.81 mg of unaltered 1,5 enyne (E)-1-(ethynyl)-2-(but-1-enyl)benzene (0.018 mmol, 13%) [\(Table 2,](#page-2-0) entry 8).

Data for 3e: colorless oil. R_f 0.19 (Hex). ¹H NMR, COSY, NOESY $(500 \text{ MHz}, \text{CDCl}_3)$ δ 7.49 (dd, J = 7.1, 3.8 Hz, 2H), 7.44 (d, J = 7.6 Hz, 1H), 7.34 (t, J = 7.4 Hz, 1H), 7.25−7.10 (m, 3H), 6.17−6.12 (m, 1H, H2′), 6.13 (t, J = 3.3 Hz, 1H, H4), 3.35−3.30 (m, 2H, H1′), 3.02−2.92 (m, 1H, H9a), 2.86 (s, 1H, H2), 2.82−2.77 (m, 2H, H9),

2.77−2.70 (m, 1H, H3), 2.38−2.30 (m, 1H, H1), 1.29 (d, J = 7.6 Hz, 3H), 1.09 (d, $J = 7.2$ Hz, 3H). NOESY 2D (500 MHz, CDCl₃): correlation peaks between H2′-H3, H3−H4. 13C{1 H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 149.3, 145.3, 145.2, 144.8, 141.2, 140.2 (C), 128.1 (C2′), 127.6, 126.5, 126.1, 125.4, 124.6, 124.1, 121.0 (C4), 120.2, 119.1, 45.8 (C2), 39.7 (CH, C9a), 37.9 (CH₂, C1'), 34.3 (CH, C3), 33.2 (CH₂, C9), 31.6 (CH₂, C1), 23.8, 16.7 (CH₃). IR (film): 3018, 2957, 2926, 2872, 1742, 1668, 1605, 1462, 1377, 1255, 1153, 1021, 969, 771, 754, 724 cm[−]¹ . HRMS (MALDI/TOF) m/z calcd for $C_{24}H_{23}$ [M + H]⁺ 311.1799; found 311.1784.

 $3-(2$ -Methylprop-1-en-1-yl)-1H-indene (4a). 8 ¹H NMR (300 MHz, CDCl₃) δ 7.48 (d, J = 7.3 Hz, 1H), 7.40 (d, J = 7.4 Hz, 1H), 7.31 (t, J $= 7.4$ Hz, 1H), 7.21 (td, $J = 7.3$, 1.3 Hz, 1H), 6.34 (s, 1H), 6.19 (m, 1H), 3.46 (s, 2H), 1.97 (s, 3H), 1.91 (s, 3H). HRMS (ESI) m/z calcd for $C_{13}H_{15}$ $[M + H]^+$ 171.1168; found 171.1170.

1,3-Dimethylnaphthalene (5) .¹¹ Following the general protocol, AgSbF₆ (5.73 mg, 16.7 μ mol, 5 mol %), Ph₃PAuCl (8.25 mg, 16.7 μ mol, 5 mol %) and enyne 1d (52.1 mg, 0.334 mmol) were mixed in 5.2 mL of CH_2Cl_2 and protected from light. After filtration through Celite and purification (Hex), 33.0 mg of 5 (0.212 mmol, 63%) were obtained. ¹H NMR (300 MHz, CDCl₃) δ 7.99–7.90 (m, 1H), 7.80– 7.72 (m, 1H), 7.51−7.39 (m, 3H), 7.18 (s, 1H), 2.67 (s, 3H), 2.48 (s, 3H). MS (EI) m/z 156 (M⁺, 100), 141 (92), 128 (14), 115 (18).

Procedure for the Cross-Cascade between 1a and 1b. Following the general protocol, to a mixture of $SS-L_1$ -AuCl (3.11 mg, 5.6 μ mol, 3 mol %) and NaBArF (4.94 mg, 5.6 μ mol, 3 mol %) in 1.7 mL of $CH₂Cl₂$ was added a solution of 1a (16.3 mg, 96 μ mol, 0.53 equiv) and 1b (18.3 mg, 91 μ mol, 0.50 equiv) in 1.7 mL of CH₂Cl₂. After 1 h, analysis by ¹H NMR of the reaction crude indicated an equimolar mixture of $2a:2b:2ab:2ba$. Purification (Hex:CH₂Cl₂) gave 5.7 mg of 2a (0.017 mmol, 18%), 7.2 mg of 2ab (0.019 mmol, 21%), 6.7 mg of 2ba (0.018 mmol, 20%), and 7.4 mg of 2b (0.019 mmol, 21%).

Data for 2-(5-Methoxy-1H-inden-3-yl)-1,1,3,3-tetramethyl-2,3,4,9-tetrahydro-1H-fluorene (2ab). Colorless oil. R_f 0.39 $(Hex:CH₂Cl₂, 2:1).$ ¹H NMR, COSY (500 MHz, CDCl₃) δ 7.44 (d, $J = 7.3$ Hz, 1H), 7.37 (d, $J = 8.1$ Hz, 1H, H7'), 7.30 (t, $J = 7.3$ Hz, 1H), 7.25 (d, J = 6.1 Hz, 1H), 7.17 (td, J = 7.3, 1.2 Hz, 1H), 6.96 (d, J = 2.3 Hz, 1H, H4'), 6.77 (dd, $J = 8.1$, 2.3 Hz, 1H, H6'), 6.45 (t, $J = 1.8$ Hz, 1H, H2′), 3.84 (s, 3H, OMe), 3.41 (s, 4H, H9+H1′), 3.02 (s, 1H, H2), 2.47 (dt, J = 16.6, 2.1 Hz, 1H, H4), 2.41 (dt, J = 16.6, 3.3 Hz, 1H, H4), 1.30 (s, 3H), 1.16 (s, 3H), 1.11 (s, 3H), 0.93 (s, 3H). ¹³C{¹H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 158.9, 150.53, 148.6, 146.4, 143.2, 143.0, 135.5, 133.4 (C), 131.94 (C2′), 126.3, 124.1, 123.9 (C7'), 123.6, 118.1, 109.7 (C6'), 105.6 (CH, C4'), 55.7 (CH₃), 51.8 (C2), 39.1 (CH₂), 37.8 (CH₂), 37.5 (C), 36.5 (CH₂), 35.1 (C), 32.5, 31.2, 25.2, 24.1 (CH3). IR (film): 3046, 2961, 2888, 2833, 1618, 1607, 1573, 1472, 1396, 1381, 1363, 1289, 1265, 1217, 1180, 1166, 1147, 1052, 1033, 798, 759, 739, 722, 705 cm[−]¹ . HRMS (ESI) m/z calcd for $C_{27}H_{31}O$ [M + H]⁺: 371.2369; found 371.2363.

Data for 2-(1H-Inden-3-yl)-6-methoxy-1,1,3,3-tetramethyl-2,3,4,9-tetrahydro-1H-fluorene (2ba). Colorless oil. R_f 0.35 $(Hex:CH₂Cl₂, 2:1).$ ¹H NMR (300 MHz, CDCl₃) δ 7.49 (d, J = 7.3 Hz, 1H), 7.41 (d, J = 7.6 Hz, 1H), 7.31 (d, J = 7.7 Hz, 2H), 7.21 (t, J = 7.1 Hz, 1H), 6.82 (d, J = 2.4 Hz, 1H, H5), 6.72 (dd, J = 8.0, 2.4 Hz, 1H, H7), 6.42 (t, J = 2.0 Hz, 1H, H2'), 3.86 (s, 3H, OMe), 3.46 (s, 2H, H1'), 3.35 (t, J = 2.4 Hz, 2H, H9), 3.09 (s, 1H, H2), 2.41 (s, 2H, H4), 1.30 (s, 3H), 1.16 (s, 3H), 1.09 (s, 3H), 0.91 (s, 3H). $^{13}C(^{1}H)$ NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 159.2, 150.2, 148.9, 147.9, 143.3, 143.2, 135.3, 133.3 (C), 130.4 (C2′), 126.1, 124.5, 123.9, 123.7, 119.2, 109.7 (C7), 104.2 (C5), 55.7 (CH₃), 51.7 (CH, C2), 39.1 (CH₂, C1'), 38.6 (CH₂), 37.6 (C), 35.8 (CH₂), 35.1 (C), 32.4, 31.2, 25.2, 24.1 (CH3). IR (film): 2962, 2887, 2833, 1618, 1607, 1574, 1475, 1381, 1287, 1265, 1214, 1163, 1033, 910, 858, 782, 760, 738, 722, 705 cm⁻¹. HRMS (ESI) m/z calcd for C₂₇H₃₁O [M + H]⁺: 371.2369; found 371.2381.

General Procedure for the [2+2+2] Cycloisomerization-**Cross-Cyclization Cascades.** A mixture of the $gold(I)$ catalyst (3% mol), the activating agent (3% mol), and benzaldehyde (5 equiv) in 1 mL of anhydrous CH_2Cl_2 was prepared in a 2-necked roundbottom flask under argon at rt. A 0.08−0.12 M solution of the enyne in anhydrous CH_2Cl_2 (1 equiv, 0.12 mmol) was slowly added (1 mL/ h) by a syringe pump. After completion of the addition, the crude was filtered through a short pad of Celite and purified by flash chromatography on silica gel.

(1SR,9aSR)-3,3-Dimethyl-1-phenyl-1,3,9,9a-tetrahydroindeno- [2,1-c]pyran (6a). Following the general protocol, 1a $(19.2 \text{ mg}, 0.113)$ mmol) in 1 mL of CH_2Cl_2 was slowly added $(1 mL/h)$ to a mixture of SS-L₁-AuCl (1.89 mg, 3.4 μ mol, 3 mol %), NaBArF (3.00 mg, 3.4 μ mol, 3 mol %), and benzaldehyde (57.4 μ L, 0.56 mmol, 5 equiv) in CH₂Cl₂. Purification (Hex:CH₂Cl₂) afforded 21.9 mg of 6a (0.079 mmol, 70%) and 3.1 mg 3a (0.009 mmol, 16%) ([Table 3,](#page-3-0) entry 1).

Data for 6a: crystalline solid. Mp 64-66 °C. R_f 0.23 $(Hex:CH_2Cl_2, 10:1)$. ¹H NMR, COSY, NOESY (500 MHz, CDCl₃) δ 7.48−7.41 (m, 3H), 7.39 (t, J = 7.5 Hz, 2H), 7.32 (t, J = 7.3 Hz, 1H), 7.24−7.16 (m, 3H), 5.96 (d, J = 2.8 Hz, 1H, H4), 4.48 (d, J = 9.5 Hz, 1H, H1), 3.09 (dddd, J = 9.5, 8.7, 8.6, 2.8 Hz, 1H, H9a), 2.78 (dd, J = 15.7, 8.6 Hz, 1H, H9 α), 2.64 (dd, J = 15.7, 8.7 Hz, 1H, H9 β), 1.48 (s, 3H), 1.41 (s, 3H). NOESY 2D (400 MHz): correlation peaks between H9a-H9, H1-H-9, H1-Me, H1-ArH. ¹³C{¹H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 144.5, 141.7, 140.9, 140.4 (C), 128.7 (Ph- $2C_{meta}$), 128.3, 128.0, 127.1 (Ph- $2C_{ortho}$), 126.9, 125.3, 124.3 (C4), 120.7, 77.6 (CH, C1), 74.6 (C), 46.7 (CH, C9a), 33.9 (CH₂, C9), 29.9, 27.2 (CH₃). IR (KBr): 3034, 2970, 2923, 1604, 1462, 1358, 1230, 1157, 1039, 750, 701 cm⁻¹. HRMS (ESI) *m/z* calcd for C₂₀H₂₁O $[M + H]^+$ 277.1587; found 277.1594. $[\alpha]_D^{25} \sim 0$ ($c = 1.0$, CHCl₃).

(1SR,9aRS)-3,3-Dimethyl-1-phenyl-1,3,9,9a-tetrahydroindeno- [2,1-c]pyran (7a). Following the general protocol enyne 1a (20.7 mg) 0.122 mmol), Ph₃PAuCl (1.80 mg, 3.7 μ mol, 3 mol %), NaBArF (3.24 mg, 3.7 μ mol, 3 mol %), and benzaldehyde (18.6 μ L, 0.183 mmol, 1.5 equiv) in 2 mL of CH_2Cl_2 gave after purification (Hex:EtOAc, 99:1) 7.1 mg of 6a (0.026 mmol, 21%) and 4.4 mg of 7a (0.016 mmol, 13%) [\(Table 3](#page-3-0), entry 2). 4a (50%) and 3a (8%) were also identified in the reaction mixture by ¹H NMR.

Data for 7a: crystalline solid. Mp 75−77 °C. R_f 0.09 (Hex:EtOAc, 99:1). ¹H NMR, COSY, NOESY (500 MHz, CDCl₃) δ 7.49 (d, J = 6.8 Hz, 1H), 7.28 (dd, J = 7.8, 1.9 Hz, 2H, Ph-H_{ortho}), 7.25–7.16 (m, 6H), 6.12 (d, J = 3.3 Hz, 1H, H4), 5.37 (d, J = 7.3 Hz, 1H, H1), 3.50 (dddd, $J = 9.3, 8.2, 7.3, 3.3 Hz, 1H, H9a$, 2.92 (dd, $J = 16.2, 9.3 Hz, 1H,$ H9 α), 2.62 (dd, J = 16.2, 8.2 Hz, 1H, H9 β), 1.41 (s, 3H), 1.08 (s, 3H). NOESY 2D (400 MHz): correlation peaks between H9a-H9, H1-H-9a, H1-Me, H1-ArH. ¹³C{¹H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 144.5, 140.8, 140.1, 139.8 (C), 128.9 (Ph-2C_{ortho}), 128.5, 128.2 (Ph-2C_{meta}), 127.5, 125.7, 124.4, 124.3 (C4), 120.7, 75.8 (CH, C1), 72.5 (C), 41.2 (CH, C9a), 34.4 (CH₂, C9), 29.5, 29.2 (CH₃). IR (KBr): 3018, 2961, 1602, 1461, 1356, 1269, 1155, 1032, 754, 700 cm⁻¹. MS (EI) *m/z* 276 (M⁺, 6), 261 (75), 170 (100), 155 (81).

(1SR,9aSR)-6-Methoxy-3,3-dimethyl-1-phenyl-1,3,9,9atetrahydroindeno[2,1-c]pyran (6b). Following the general protocol enyne 1b (22.7 mg, 0.113 mmol), SS-L₁-AuCl (1.90 mg, 3.4 μ mol, 3 mol %), NaBArF (3.01 mg, 3.4 μmol, 3 mol %) and benzaldehyde (57.7 μ L, 0.57 mmol, 5 equiv) in 2 mL of CH_2Cl_2 gave after purification (Hex: CH_2Cl_2) 24.8 mg of 6b (0.081 mmol, 72%) and 3.3 mg 3b (0.008 mmol, 14%) ([Table 3](#page-3-0), entry 3).

Data for 6b: colorless oil. R_f 0.18 (Hex: CH_2Cl_2 , 1:1). ¹H NMR, COSY, NOESY (500 MHz, CDCl₃) δ 7.45 (d, J = 7.3 Hz, 2H, H_{ortho}), 7.38 (t, J = 7.3 Hz, 2H, H_{meta}), 7.30 (t, J = 7.3 Hz, 1H, H_{para}), 7.09 (d, J $= 8.2$ Hz, 1H, H8), 6.95 (d, J = 2.5 Hz, 1H, H5), 6.76 (dd, J = 8.2, 2.5) Hz, 1H, H7), 5.94 (d, J = 2.9 Hz, 1H, H4), 4.47 (d, J = 9.5 Hz, 1H, H1), 3.82 (s, 3H), 3.10 (dddd, J = 9.5, 8.6, 8.5, 2.9 Hz, 1H, H9a), 2.71 $(dd, J = 15.3, 8.6 Hz, 1H, H9\alpha), 2.57 (dd, J = 15.3, 8.5 Hz, 1H, H9\beta),$ 1.48 (s, 3H), 1.40 (s, 3H). NOESY 2D (400 MHz): correlation peaks between H9a-H9, H1-H-9, H1-Me, H1-ArH. ¹³C{¹H}NMR, HSQC, HMBC (126 MHz, CDCl₃) δ 159.2, 141.7 (2C), 141.1, 136.7 (C), 128.7 (Ph–C_{meta}), 128.0 (Ph-2C_{para}), 127.1 (Ph-2C_{ortho}), 125.9 (C8), 124.3 (C4), 115.2 (C7), 105.1 (C5), 77.5 (CH, C1), 74.6 (C), 55.7 $(CH₃)$, 47.4 (CH, C9a), 33.1 (CH₂, C9), 29.9, 27.2 (CH₃). IR (film): 2969, 2932, 2855, 1679, 1610, 1487, 1466, 1454, 1358, 1328, 1285, 1238, 1176, 1147, 1032, 840, 745, 700 cm[−]¹ . HRMS (ESI) m/z calcd for $C_{21}H_{23}O_2$ [M + H]⁺ 307.1698; found 307.1750.

Crystal Data. Data were collected at low temperature using oilcoated shock-cooled crystals on a Bruker-AXS APEX II diffractometer with Mo K α radiation ($\lambda = 0.71073$ Å). The structure was solved by direct methods, 17 17 17 and all nonhydrogen atoms were refined anisotropically using the least-squares method on $F^{2.18}$ $F^{2.18}$ $F^{2.18}$.

Crystal data for 2a: Crystals were obtained by slow evaporation of a concentrated solution of compound 2a in hexanes. $C_{26}H_{28}$, M_r = 340.48, crystal dimensions $0.4 \times 0.3 \times 0.3$ mm³, monoclinic, $P2₁/c$, $a =$ 8.125(1) Å, $b = 21.014(2)$ Å, $c = 11.266(2)$ Å, $\beta = 99.032(3)$ °, cell volume = 1899.6(3) Å³, Z = 4, $\rho_{\text{calcd}} = 1.191 \text{ Mg/m}^3$, $\mu = 0.067 \text{ mm}^{-1}$, $T = 100(2)$ K, $2\theta_{\text{max}} = 52.7^{\circ}$, 31344 reflections collected, 3861 independent, $R_{\text{int}} = 0.1211$, $R1 = 0.0496$ and $wR2 = 0.1002$ for $I >$ $2\sigma(I)$, R1 = 0.0961 and wR2 = 0.1174 for all data, residual electron density = 0.244 eA⁻³. .

■ ASSOCIATED CONTENT

6 Supporting Information

The Supporting Information is available free of charge on the [ACS Publications website](http://pubs.acs.org) at DOI: [10.1021/acs.joc.7b01273.](http://pubs.acs.org/doi/abs/10.1021/acs.joc.7b01273)

> Copies of 1D and 2D NMR spectra of compounds 2, 3, 6 and 7 and ¹ H NMR of 4a and 5 and HPLC profiles for 3a and 6a ([PDF](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b01273/suppl_file/jo7b01273_si_001.pdf)) Crystal structure of 2a ([CIF\)](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b01273/suppl_file/jo7b01273_si_002.cif)

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Notes

The authors declare no competing financial interest.

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